

# Activation of carbonaceous materials by combined KOH chemical treatment and microwave irradiation

## Introduction

Activated carbon (AC) ·· High surface area  
Microporous material

Various applications  
Separation, Purification, Drying, Catalyst support  
Gas storage (H<sub>2</sub>, CH<sub>4</sub> etc...)  
Electrode for electric double layer capacity  
Adsorption heat pump, Desiccant humidity conditioner

- ◆ Pore structure (Surface area, Pore volume, Pore size)
- ◆ Surface characteristics (Surface functional group)

Be carefully controlled to improve performance

### Control factor

Carbonaceous materials : Phenol resin, Coconut shell, Coke, Plastics ...  
Activation methods : Chemical activation (KOH, NaOH, H<sub>3</sub>PO<sub>4</sub> ...)  
Gas activation (Water vapor, CO<sub>2</sub> ...)  
Activation conditions : Temperature, Heating rate, Amount of activating agent ...  
Heating methods : Electric furnace

Many researches

### Novel heating method ·· Microwave heating

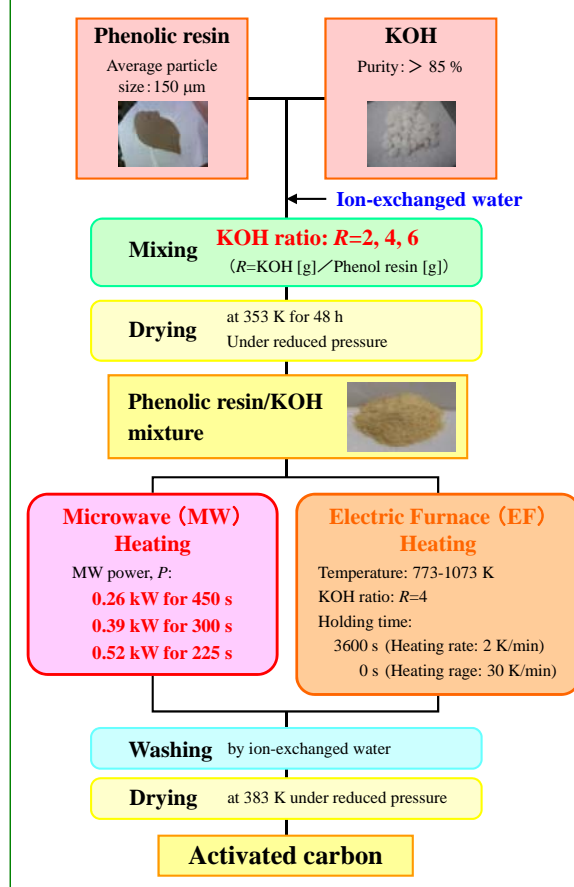
- Rapid heating
- Direct heating

### In this study

- Activated carbons were prepared by combined KOH chemical treatment and MW irradiation with various weight ratios of KOH to raw material and MW powers.
- Pore structure of activated carbon was compared for microwave and electric furnace heating.
- The applicability of activated carbon to adsorption heat pump (AHP) and desiccant humidity conditioner (DHC) was evaluated based on the effective water adsorptivity for both systems.

## Experimental

### Procedure



### Apparatus

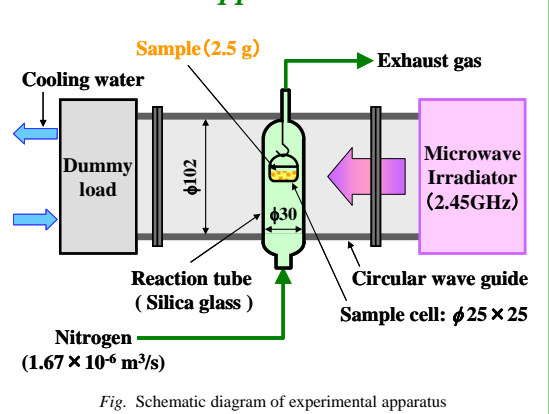


Fig. Schematic diagram of experimental apparatus

### Analysis

#### ○ Nitrogen adsorption isotherm

- BET surface area ( $S_{BET}$ )
- Pore volume: Total ( $V_{Total}$ ) at  $p/p_s=0.995$   
Micro ( $V_{Micro}$ ) at  $p/p_s=0.1$   
Meso ( $V_{Meso}$ ) by  $V_{Total} - V_{Micro}$
- Pore size distribution by BJH method

#### ○ Water vapor adsorption isotherm

- Water adsorptivity
- Effective water adsorptivity in the operation relative pressure ranges of AHP & DHC ( $\Delta q_{AHP}$ ,  $\Delta q_{DHC}$ )

## Results & Discussion

### Heating characteristics with microwave irradiation

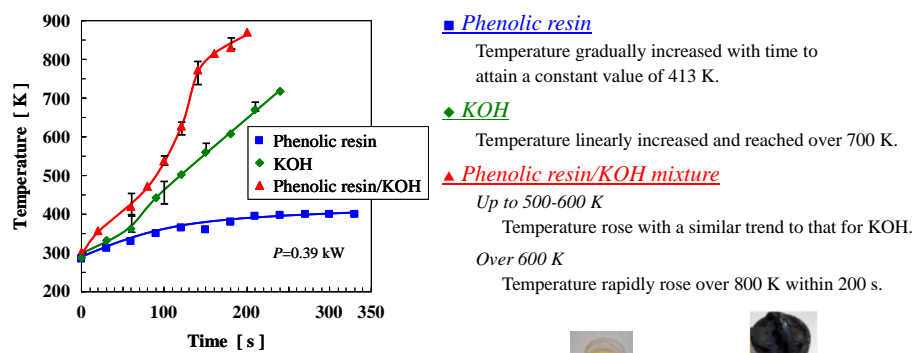


Fig. Temperature profile of samples with microwave heating

#### Initial period of MW heating (within 100 s)

Phenolic resin/KOH mixture was mainly heated by absorption of MW energy by KOH.

#### Beyond 100 s

Owing to carbonation and activation of mixture, MW absorptivity of the mixture changed significantly, which brought about a rapid temperature increase.

### Pore structure of ACs prepared with MW and with EF heating

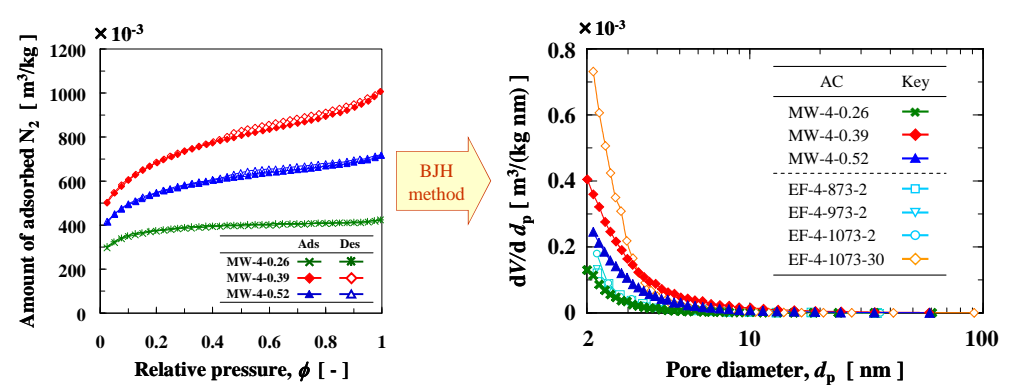


Fig. N<sub>2</sub> adsorption isotherm of ACs by microwave heating

Fig. Pore size distribution of ACs prepared with microwave and with electric furnace heating

Table Pore structure and effective water adsorptivity of ACs prepared

| a) Microwave heating |   |      |               |               |               |                      |                  | b) Electric furnace heating |              |         |      |                      |                      |                      |                      |                  |                  |
|----------------------|---|------|---------------|---------------|---------------|----------------------|------------------|-----------------------------|--------------|---------|------|----------------------|----------------------|----------------------|----------------------|------------------|------------------|
| Sample               | R | P    | $S_{BET}$     | $V_{Total}$   | $V_{Meso}$    | $V_{Meso}/V_{Total}$ | $\Delta q_{AHP}$ | $\Delta q_{DHC}$            | Sample       | r       | T    | $S_{BET}$            | $V_{Total}$          | $V_{Meso}$           | $V_{Meso}/V_{Total}$ | $\Delta q_{AHP}$ | $\Delta q_{DHC}$ |
|                      |   |      | $\times 10^3$ | $\times 10^3$ | $\times 10^3$ | [%]                  | [kg/kg]          | [kg/kg]                     |              | [K/min] | [K]  | [m <sup>2</sup> /kg] | [m <sup>3</sup> /kg] | [m <sup>3</sup> /kg] | [%]                  | [kg/kg]          | [kg/kg]          |
| MW-4-0.26            | 4 | 0.26 | 1129          | 0.656         | 0.117         | 17.8                 | 0.058            | 0.457                       | EF-4-873-2   | 2       | 873  | 1892                 | 1.083                | 0.257                | 23.7                 | 0.055            | 0.691            |
| MW-4-0.39            | 4 | 0.39 | 2208          | 1.559         | 0.622         | 39.9                 | 0.085            | 0.396                       | EF-4-973-2   | 2       | 973  | 2118                 | 1.184                | 0.268                | 22.6                 | 0.017            | 0.582            |
| MW-4-0.52            | 4 | 0.52 | 1723          | 1.114         | 0.351         | 31.5                 | 0.097            | 0.384                       | EF-4-1073-2  | 2       | 1073 | 2279                 | 1.274                | 0.315                | 24.7                 | 0.008            | 0.265            |
| MW-2-0.39            | 2 | 0.39 | 750           | 0.459         | 0.084         | 18.4                 | 0.119            | 0.325                       | EF-4-1073-30 | 30      | 1073 | 2599                 | 1.465                | 0.511                | 34.9                 | 0.006            | 0.049            |
| MW-6-0.39            | 6 | 0.39 | 1362          | 0.807         | 0.194         | 24.0                 | 0.102            | 0.366                       |              |         |      |                      |                      |                      |                      |                  |                  |

#### Microwave heating

MW-4-0.39 : Maximum  $S_{BET}$  &  $V_{Total}$  of  $2,208 \times 10^3$  m<sup>2</sup>/kg &  $1,559 \times 10^3$  m<sup>3</sup>/kg.  
MW-4-0.39 & MW-4-0.52 : Mesopores accounted for over 30 % of total pores.

Higher ratio of mesopore than conventional ACs at  $r=2$  K/min.

#### Electric furnace heating

Heating rate of 2 K/min : Microporosity.

Heating rate of 30 K/min : Mesopore ratio higher by 10 % than AC prepared at  $r=2$  K/min.

Rapid heating promoted development of mesopores

### Water vapor adsorption of ACs prepared

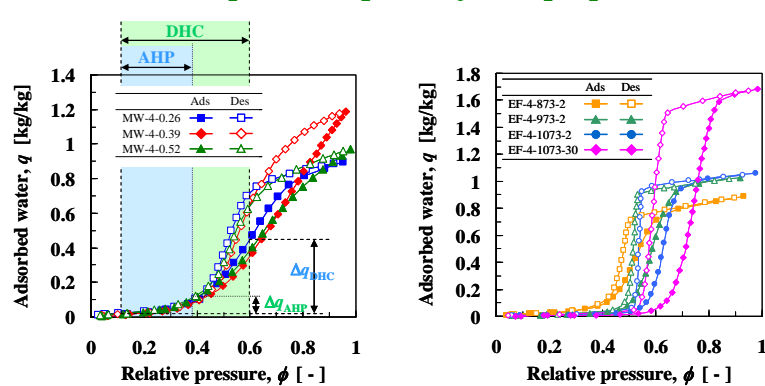


Fig. Water vapor adsorption isotherm of activated carbon

#### Microwave heating: Type III (BDDT classification)

Amount of adsorbed water gradually and monotonically increased as relative pressure.

#### Electric furnace heating: Type IV

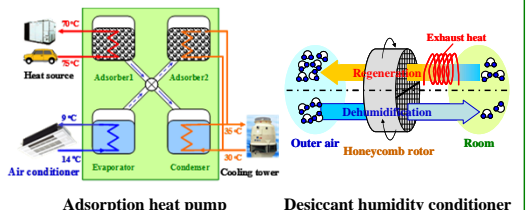
Adsorption isotherm had almost no adsorption up to around  $p/p_s=0.4$ . A steep increase in uptake was observed at higher relative pressure. (Most typical shape of water adsorption for AC).

### Applicability of AC by microwave heating to AHP & DHC

AHP :  $\Delta q_{AHP-MW-AC} \ll \Delta q_{AHP-Silica\ gel} = 0.158$  g-H<sub>2</sub>O/g-silica gel

DHC :  $\Delta q_{DHC-MW-4-0.26} = 0.457$  g/g >  $\Delta q_{DHC-Silica\ gel} = 0.295$  g/g

Activated carbons by microwave heating have an applicability to desiccant humidity conditioner



## Conclusion

- ◆ Both phenolic resin/KOH mixture and potassium hydroxide were rapidly heated over 700 K within 240 s, activated carbon was produced from the mixture in a short time.
- ◆ For AC prepared at KOH/ phenolic resin weight ratio, R, of 4 and at MW power, P, of 0.39 kW,  $S_{BET}$  and  $V_{Total}$  reached maximum values of  $2,208 \times 10^3$  m<sup>2</sup>/kg and  $1,559 \times 10^3$  m<sup>3</sup>/kg, respectively.
- ◆ Activated carbon prepared under microwave heating at  $P=0.39$  and 0.52 kW had high ratio of mesopore to total pore compared to ACs prepared with slow electric furnace heating.
- ◆ Activated carbon prepared prepared at  $R=4$  and at  $P=0.26$  kW had 1.5 times higher effective water adsorptivity for desiccant humidity conditioner than a commercial silica gel.

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